

SCHEME OF INSTRUCTION AND SYLLABI (R-20)

OF

B.E. I & II SEMESTERS

IN

CIVIL ENGINEERING

(For the batch admitted in 2020-21)



CHAITANYA BHARATHI INSTITUTE OF TECHNOLOGY

(An Autonomous Institution)
Affiliated to Osmania University

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CHAITANYA BHARATHI INSTITUTE OF TECHNOLOGY (AUTONOMOUS)

DEPARTMENT OF CIVIL ENGINEERING

INSTITUTE VISION AND MISSION

Vision

To be a centre of excellence in technical education and research.

Mission

To address the emerging needs through quality technical education and advanced research.

DEPARTMENT VISION AND MISSION

Vision

To strive for excellence in academics, research and consultancy in the field of Civil Engineering and contribute to the sustainable development of the country by producing quality Civil Engineers with professional and ethical values.

Mission

- Maintaining high academic standards to develop analytical thinking and independent judgment among the students so that they are fit for industry and higher studies.
- Promoting skills and values among the students to prepare them as responsible global citizens who can solve complex problems.
- Preparing the students as good individuals and team members with professional attitude, ethics, concern for environment and zeal for lifelong learning who can contribute to society.

PROGRAM EDUCATIONAL OBJECTIVES:

The PEOs are to facilitate the graduating students to

PEO1: Acquire basic knowledge and expertise necessary for professional practice in Civil Engineering for higher studies and research.

PEO2: Attain and practice technical skills to identify, analyze and solve complex problems and issues related to Civil Engineering.

PEO3: Possess a professional attitude as an individual or a team member to work for the betterment of the society and environment.

PEO4: Work with professional ethics as refined technocrats with a thirst for lifelong learning.

PROGRAM SPECIFIC OUTCOMES:

The graduates of this program will:

- 1. Effectively apply engineering fundamentals for the development and management of ecofriendly Civil engineering systems which benefit the society at large.
- 2. Develop the ability to provide solutions to complex problems in civil engineering through individual and team work with a spirit for lifelong learning
- 3. Develop the competence to plan, build and maintain sustainable infrastructural facilities like housing, water management, transportation and geotechnical services.

PROGRAM OUTCOMES:

Engineering graduate will be able to:

1. Engineering Knowledge	Apply the knowledge of mathematics, science, engineering fundamentals, and an engineering specialization to the solution of complex engineering problems
2. Problem Analysis	Identify, formulate, review of research literature, and analyse complex engineering problems reaching substantiated conclusions using first principles of mathematics, natural sciences, and engineering sciences.
3. Design/ development of Solutions	Design solutions for complex engineering problems and design system components or processes that meet the specified needs with appropriate consideration for public health and safety, and cultural, societal, and environmental considerations.
4. Conduct Investigations of Complex Problems	Use research-based knowledge and research methods including design of experiments, analysis and interpretation of data, and synthesis of the information to provide valid conclusions.
5. Modern Tool Usage	Create, select, and apply appropriate techniques, resources, and modern engineering and IT tools, including prediction and modelling to complex engineering activities, with an understanding of the limitations.
6. The Engineer and Society	Apply reasoning informed by the contextual knowledge to assess societal, health, safety, legal and cultural issues and the consequent responsibilities relevant to the professional engineering practice.
7. Environment and Sustainability	Understand the impact of the professional engineering solutions in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.
8. Ethics	Apply ethical principles and commit to professional ethics and responsibilities and norms of the engineering practice.
9. Individual and Team Work	Function effectively as an individual, and as a member or leader in diverse teams, and in multidisciplinary settings. Communicate effectively on complex engineering activities with the
10. Communication	engineering community and with the society at large, such as, being able to comprehend and write effective reports and design documentation, make effective presentations, and give and receive clear instructions.
11. Project Management and Finance	Demonstrate knowledge and understanding of the engineering and management principles and apply these to one's own work, as a member and leader in a team, to manage projects and in multidisciplinary environments.

About the Department:

change.

12. Life-Long

Learning

The Department of Civil Engineering was established in the Year 1979 with the inception of the Institute. The Department offers a Four-Year Bachelor's Degree Program in Civil Engineering and Two-Year Master's Degree Program in Structural Engineering. In total there are about 500 Students pursuing Bachelor's Degree and 50 Students pursuing Master's Degree in the Department. Many of the Faculty Members hold Higher Degrees from reputed Institutions from India like IITs, NITs, IIITH and State Universities and Abroad. The Department of Civil Engineering has Three missions such as excellence in Teaching and Research; Relevance to Industry and Society; contribution to sustainable development. The Department is well equipped with State-of-the-Art laboratories to cater the needs of Teaching, R&D and Consultancy. The Faculty of the Department strive hard to impart the latest technical knowledge to the students and conduct quality research. The faculty offer technical services on live engineering problems to various Government and Private organizations.

Recognise the need for, and have the preparation and ability to engage in

independent and life-long learning in the broadest context of technological



CHAITANYA BHARATHI INSTITUTE OF TECHNOLOGY (AUTONOMOUS)

Scheme of Instructions of I Semester of B.E. –Civil Engineering as per AICTE Model Curriculum 2020-21

DEPARTMENT OF CIVIL ENGINEERING

SEMESTER - I

G	Course Code		Scheme of Instruction			Scheme of Examination				
S. No		Title of the Course	Hours per Week			Duration of SEE	Maximum Marks		Credits	
			L	T	P/D	in Hours	CIE	SEE		
		T	HEOR	Y						
1	20MT C05	Calculus	3	1	-	3	40	60	4	
2	20CYC01	Chemistry	3	ı	-	3	40	60	3	
3	20CE C01	Engineering Mechanics-I	3	-	-	3	40	60	3	
4	20CS C01	Programming for Problem Solving	3	-	-	3	40	60	3	
	PRACTICAL									
5	20CYC02	Chemistry Lab	-	-	4	3	50	50	2	
6	20CS C02	Programming for Problem Solving Lab	-	-	4	3	50	50	2	
7	20ME C02	Workshop/ Manufacturing Practice	-	-	5	3	50	50	2.5	
8	20ME C03	TE C03 Engineering Exploration		Hours	/ 4P	-	50	-	1.5	
	TOTAL			1	13	-	360	390	21	

L: Lecture T: Tutorial P: Practical

CIE - Continuous Internal Evaluation SEE - Semester End Examination

20MT C05

CALCULUS (Common to ECE, EEE, MECH, CHEM, CIVIL)

Instruction 3 L+1T/2P Hours per week
Duration of SEE 3 Hours
SEE 60 Marks
CIE 40 Marks
Credits 4

Course Objectives:

- 1. To explain the solutions of system of linear equations by Matrix Methods.
- 2. To discuss mean value theorems.
- 3. To explain the Partial Derivatives and the extreme values of functions of two variables.
- 4. To discuss the convergence and divergence of the series.
- 5. To explain the expansion of functions in sine and cosine series.

Course Outcomes:

Upon completing this course, students will be able to:

- 1. Apply the Matrix Methods to solve system of linear equations.
- 2. Analyse the geometrical interpretation of Mean value theorems.
- 3. Determine the extreme values of functions of two variables.
- 4. Examine the convergence and divergence of infinite Series.
- 5. Calculate the Euler's coefficients for Fourier series of a function.

UNIT-I

Matrices: Rank of a matrix, Echelon form, consistency of linear system of equations, Linear dependence and independence of vectors. Eigen values, Eigenvectors, Properties of Eigen values& Eigen vectors, Cayley-Hamilton theorem, Quadratic form, Reduction of quadratic form to canonical form by linear transformation, Nature of quadratic form.

UNIT-II

Calculus: Rolle's Theorem, Lagrange's Mean value theorem, Cauchy's mean value theorem (without proofs). Curvature, Radius of curvature, Centre of curvature, Evolute and Involute.

UNIT-III

Multivariable Calculus (Differentiation): Functions of two variables, Partial derivatives, Higher order partial derivatives, Total derivative, Differentiation of implicit functions, Change of variables, Jacobians, Taylor's theorem for functions of two variables, Maxima and minima of functions of two variables.

UNIT-IV

Sequences and Series: Convergence of sequence and series. Tests for convergence of series: Comparison test, limit comparison test, D'Alembert's ratio test, Raabe's test, Cauchy's root test, alternating series, Leibnitz's series, absolute and conditional convergence.

UNIT-V

Fourier series: Periodic functions, Euler' formulae, Conditions for a Fourier expansion, functions having points of discontinuity, change of interval, even and odd functions, half range sine series, half range cosine series, and its applications in practical Harmonic analysis.

Text Books:

- 1. B.S.Grewal, Higher Engineering Mathematics, Khanna Publishers, 44th Edition, 2017.
- 2. Erwin Kreyszig, Advanced Engineering Mathematics, 9th Edition, John Wiley & Sons, 2006.
- 3. B.V.Ramana., Higher Engineering Mathematics, Tata McGraw-Hill, New Delhi, 11th Reprint, 2010.

- 1. Veerarajan T., Engineering Mathematics for first year, Tata McGraw-Hill, New Delhi, 2008.
- 2. R.K.Jain, S.R.K.Iyengar, Advanced Engineering Mathematics, Narosa Publications, 5th edition, 2016.
- 3. David.Poole, Linear Algebra: A Modern Introduction, 2nd Edition, Brooks/ Cole, 2005.

20CY C01

CHEMISTRY

(Common to all branches)

Instruction: 3Hours per Week

Duration of SEE: 3 Hours
SEE: 60 Marks
CIE: 40 Marks
Credits: 3

Course Objectives

- 1. This syllabus helps at providing the concepts of chemical bonding and chemical kinetics to the students aspiring to become practicing engineers
- 2. Thermodynamic and Electrochemistry units give conceptual knowledge about processes and how they can be producing electrical energy and efficiency of systems.
- 3. To teach students the value of chemistry and to improve the research opportunities knowledge of stereochemistry and organic reactions is essential.
- 4. Water chemistry unit impart the knowledge and understand the role of chemistry in the daily life.
- 5. New materials lead to discovering of technologies in strategic areas for which an insight into Polymers, nanomaterials and basic drugs of modern chemistry is essential.

Course Outcomes

At the end of the course student will be able to:

- 1. Identify the microscopic chemistry in terms of molecular orbitals, intermolecular forces and rate of chemical reactions.
- 2. Discuss the properties and processes using thermodynamic functions, electrochemical cells and their role in batteries and fuel cells.
- 3. Illustrate the major chemical reactions that are used in the synthesis of organic molecules.
- 4. Classify the various methods used in treatment of water for domestic and industrial use.
- 5. Outline the synthesis of various Engineering materials & Drugs.

UNIT-I Atomic and molecular structure and Chemical Kinetics:

Atomic and molecular structure: Molecular Orbital theory - atomic and molecular orbitals. Linear combination of atomic orbitals (LCAO) method. Molecular orbitals of diatomic molecules. Molecular Orbital Energy level diagrams (MOED) of diatomic molecules & molecular ions (H_2 , He_2^+ , N_2 , O_2 , O_2^- , O_3 , O_3 . Pi- molecular orbitals of benzene and its aromaticity.

Chemical Kinetics: Introduction, Terms involved in kinetics: rate of reaction, order & molecularity; First order reaction-Characteristics: units of first order rate constant & its half-life period, second order reaction-Characteristics: units of second order rate constant & its half-life period. Numericals.

UNIT-II Use of free energy in chemical equilibria

Use of free energy in chemical equilibria: Thermodynamic functions: Internal energy, entropy and free energy. Significance of entropy and free energy (criteria of spontaneity). Free energy and emf (Gibbs Helmholtz equations and its applications). Cell potentials, electrode potentials, – Reference electrodes (NHE, SCE)-electrochemical series. Nernst equation and its applications. Determination of pH using combined Glass & Calomel electrode. Potentiometric Acid base & Redox Titrations. Numericals.

Battery technology: Rechargeable batteries & Fuel cells.

Lithium batteries: Introduction, construction, working and applications of Li-MnO₂ and Li-ion batteries. Fuel Cells: Introduction, difference between conventional cell and fuel cell, limitations & advantages. Construction, working & applications of methanol-oxygen fuel cell.

UNIT- III Stereochemistry and Organic reactions

Stereochemistry: Representations of 3 dimensional structures, Types of stereoisomerism- Conformational isomerism – confirmations of n-butane (Newman and sawhorse representations), Configurational isomerism - Geometrical (cis-trans) isomerism & Optical isomerism- optical activity, Symmetry and chirality: Enantiomers (lactic acid) & Diastereomers (Tartaric acid), Absolute configurations, Sequence rules for R&S notation.

Types of Organic reactions: Substitution Reactions- Electrophilic substitution (Nitration of Benzene); Nucleophilic Substitution ($S_N1\&S_N2$); Free Radical Substitution (Halogenation of Alkanes)

Addition Reactions: Electrophilic Addition – Markonikoff's rule, Free radical Addition - Anti Markonikoff's rule (Peroxide effect), Nucleophilic Addition – (Addition of HCN to carbonyl compounds)

Eliminations-E1 and E2 (dehydrohalogenation of alkyl halides)

Cyclization (Diels - Alder reaction)

UNIT-IV Water Chemistry:

Hardness of water – Types, units of hardness, Disadvantages of hard water, Alkalinity and Estimation of Alkalinity of water, Boiler troubles - scales & sludge formation, causes and effects, Softening of water by lime soda process (Cold lime soda process),ion exchange method and Reverse Osmosis. Specifications of potable water & industrial water.Disinfection of water by Chlorination; break point chlorination, BOD and COD definition, Estimation (only brief procedure) and significance, Numericals.

UNIT-V Engineering Materials and Drugs:

Introduction, Terms used in polymer science; Thermoplastic polymers (PVC) &Thermosetting polymers (Bakelite); Elastomers (Natural rubber). Conducting polymers - Definition, classification and applications.

Polymers for Electronics: Polymer resists for integrated circuit fabrication, lithography and photolithography.

Nano materials-Introduction to nano materials and general applications, basic chemical methods of preparation-Sol-gel method. Carbon nanotubes and their applications. Characterisation of nanomaterials by SEM and TEM (only Principle).

Drugs-Introduction, Synthesis and uses of Aspirin (analgesic), Paracetamol (Antipyretic), Atenolol (antihypertensive).

Text Books:

- 1. P.C. Jain and M. Jain, "Engineering Chemistry", Dhanpat Rai Publishing Company Ltd., New Delhi,16th edition (2015).
- 2. W.U. Malik, G.D.Tuli and R.D.Madan, "Selected topics in Inorganic Chemistry", S Chand & Company Ltd, New Delhi, reprint (2009).
- 3. R.T. Morrison, R.N. Boyd and S.K. Bhattacharjee, "Organic Chemistry", Pearson, Delhi, 7th edition (2019).
- 4. A Textbook of Polymer Science and Technology, Shashi Chawla, Dhanpat Rai & Co. (2014)
- 5. T. Pradeep, Nano: The Essentials, Tata McGraw-Hill Education, Delhi, 2012
- 6. G.L. David Krupadanam, D. Vijaya Prasad, K. Varaprasad Rao, K.L.N. Reddy and C.Sudhakar, "Drugs", Universities Press (India) Limited, Hyderabad (2007).

- 1. B. H. Mahan, "University Chemistry", Narosa Publishing house, New Delhi, 3rd edition (2013).
- 2. B.R. Puri, L.R. Sharma and M.S. Pathania, "Principles of Physical Chemistry", S. Nagin Chand & Company Ltd., 46thedition(2013).
- 3. T.W. Graham Solomons, C.B. Fryhle and S.A. Snyder, "Organic Chemistry", Wiley,12th edition (2017).
- 4. P.W. Atkins, J.D. Paula, "Physical Chemistry", Oxford, 8th edition (2006).

20CE C01

ENGINEERING MECHANICS - I

Instruction: 3 L Hours per week
Duration of SEE: 3 Hours
SEE: 60 Marks
CIE: 40 Marks
Credits: 3

Course Outcomes: At the end of the course the student will be able to

- 1. Calculate the components and resultant of coplanar forces system.
- 2. Understand free body diagram and apply equilibrium equations to solve for unknown forces.
- 3. Apply concepts of friction for solving engineering problems.
- 4. Analyse simple trusses for forces in various members of a truss.
- 5. Determine centroid for elementary, composite figures and bodies.

UNIT-I:

Resolution and Resultant of Force System: Basic concepts of a force system. Components of forces in a plane. Resultant of coplanar concurrent force system. Moment of a force, couple and their applications. Resultant of coplanar non-concurrent force system.

UNIT - II:

Equilibrium of Force System: Free body diagram, equations of equilibrium. Lami's theorem, equilibrium of coplanar force systems.

UNIT - III:

Friction: Theory of static friction, Laws of friction, applications to single body, connected systems, wedge friction and belt friction.

UNIT - IV:

Analysis of Simple Trusses: Introduction to trusses, analysis of simple trusses using method of joints and method of sections.

UNIT - V:

Centroid and Centre of Gravity:

Centroid of lines and areas from first principles, centroid of composite figures, theorems of Pappus, centre of gravity of bodies and composite bodies.

Text Books:

- 1. K. Vijaya Kumar Reddy and J. Suresh Kumar, "Singer's Engineering Mechanics: Statics and Dynamics", B. S. Publications (SI Units), 3rd edn. Rpt., 2019.
- 2. A. Nelson., "Engineering Mechanics", Tata McGraw Hill, Delhi, 2010.

- 1. Irving H. Shames, "Engineering Mechanics", 4th Edition, Prentice Hall, 2006.
- 2. R. C. Hibbeler, "Engineering Mechanics: Principles of Statics and Dynamics", Pearson Press, 2006.
- 3. Basudeb Bhattacharyya, "Engineering Mechanics", Oxford University Press, 2nd edn., 2016.

PROGRAMMING FOR PROBLEM SOLVING (Common to all Programs)

Instruction 3 Periods per week

Duration of SEE 3Hours
SEE 60 Marks
CIE 40 Marks
Credits 3

Course Objectives: The objectives of this course are

- 1. Identification of computer components, Operating environments, IDEs.
- 2. Understanding the steps in problem solving and formulation of algorithms to problems.
- 3. Develop programming skills as a means of implementing an algorithmic solution with appropriate control and data structures.
- 4. Develop in tuition to enable students to come up with creative approaches to problems.
- 5. Manipulation of text data using files.

Course Outcomes: On Successful completion of the course, students will be able to

- 1. Identify and understand the computing environments for scientific and mathematical problems.
- Formulate solutions to problems with alternate approaches and represent them using algorithms / Flowcharts.
- 3. Choose data types and control structures to solve mathematical and scientific problem.
- 4. Decompose a problem into modules and use functions to implement the modules.
- 5. Apply arrays, pointers, structures, and unions to solve mathematical and scientific problems.
- 6. Develop applications using file I/O.

UNIT-I

Introduction to computers and Problem Solving: Components of a computer, Operating system, compilers, Program Development Environments, steps to solve problems, Algorithm, Flowchart / Pseudocode with examples.

Introduction to programming: Programming languages and generations, categorization of high-level languages.

Introduction to C: Introduction, structure of C program, keywords, identifiers, Variables, constants, I/O statements, operators, precedence, and associativity.

UNIT - II

Introduction to decision control statements: Selective, looping, and nested statements.

Functions: Introduction, uses of functions, Function definition, declaration, passing parameters to functions, recursion, scope of variables and storage classes, Case study using functions and control statements.

UNIT - III

Arrays: Introduction, declaration of arrays, accessing and storage of array elements, 1-dimensional array, Searching (linear and binary search algorithms) and sorting (Selection and Bubble) algorithms, 2-D arrays, matrix operations.

Strings: Introduction, strings representation, string operations with examples. Case study using arrays.

UNIT - IV

Pointers: Understanding computer's memory, introduction to pointers, declaration pointer variables, pointer arithmetic, pointers and strings, array of pointers, dynamic memory allocation, advantages, and drawbacks of pointers.

Structures: Structure definition, initialization and accessing the members of a structure, nested structures, structures and functions, self- referential structures, unions, and enumerated data types.

UNIT-V

Files: Introduction to files, file operations, reading data from files, writing data to files, error handing during file operations.

Preprocessor Directives: Types of preprocessor directives, examples.

Text Books:

- 1. M.T. Somashekar "Problem Solving with C", 2nd Edition, Prentice Hall India Learning Private Limited 2018
- 2. AK Sharma, "Computer Fundamentals and Programming", 2nd Edition, University Press, 2018
- 3. Pradeep Deyand Manas Ghosh, "Programming in C", Oxford Press, 2nd Edition, 2017

Suggested Reading:

- 1. Byron Gottfried, Schaum's Outline of Programming with C", Mc Graw-Hill.
- 2. Brian W. Kernighan and Dennis M. Ritchie, The C Programming Language, Prentice Hall of India.
- 3. E. Balaguruswamy, Programming in ANSI C, Tata McGraw-Hill.
- 4. Reema Tharaja "Introduction to C Programming", Second Edition, OXFORDPress, 2015.

Online Resources:

- 1. https://www.tutorialspoint.com/cprogramming/index.htm.
- 2. https://onlinecourses.nptel.ac.in/noc18-cs10/preview

20CY C02

CHEMISTRY LAB

(Common to all branches)

Instruction: 4 Hours per Week
Duration of SEE: 3 Hours
SEE: 50 Marks
CIE: 50 Marks
Credits: 2

Course Objectives

- 1. To impart fundamental knowledge in handling the equipment / glassware and chemicals in chemistry laboratory.
- 2. To provide the knowledge in both qualitative and quantitative chemical analysis
- 3. The student should be conversant with the principles of volumetric analysis
- 4. To apply various instrumental methods to analyse the chemical compounds and to improve understanding of theoretical concepts.
- 5. To interpret the theoretical concepts in the preparation of new materials like drugs and polymers.

Course Outcomes

At the end of the course student will be able to:

- 1. Identify the basic chemical methods to analyse the substances quantitatively & qualitatively.
- 2. Estimate the amount of chemical substances by volumetric analysis.
- 3. Determine the rate constants of reactions from concentration of reactants/ products as a function of time.
- 4. Calculate the concentration and amount of various substances using instrumental techniques.
- 5. Develop the basic drug molecules and polymeric compounds.

Chemistry Lab

- 1. Introduction: Preparation of standard solution of oxalic acid and standardisation of NaOH.
- 2. Estimation of metal ions (Co⁺²& Ni⁺²) by EDTA method.
- 3. Estimation of temporary and permanent hardness of water using EDTA solution
- 4. Determination of Alkalinity of water
- 5. Determination of rate constant for the reaction of hydrolysis of methyl acetate. (first order)
- 6. Determination of rate constant for the reaction between potassium per sulphate and potassium Iodide. (second order)
- 7. Estimation of amount of HCl Conductometrically using NaOH solution.
- 8. Estimation of amount of HCl and CH₃COOH present in the given mixture of acids Conductometrically using NaOH solution.
- 9. Estimation of amount of HCl Potentiometrically using NaOH solution.
- 10. Estimation of amount of Fe⁺²Potentiometrically using KMnO₄ solution
- 11. Preparation of Nitrobenzene from Benzene.
- 12. Synthesis of Aspirin drug and Paracetamol drug.
- 13. Synthesis of phenol formaldehyde resin.

Text Books:

- 1. J. Mendham and Thomas , "Vogel's text book of quantitative chemical analysis", Pearson education Pvt. Ltd. New Delhi $,6^{th}$ ed. 2002.
- 2. Senior practical physical chemistry by B.D.Khosla, V.C.Garg & A.Gulati,; R. Chand & Co. : New Delhi (2011).

- 1. Dr. Subdharani, "Laboratory Manual on Engineering Chemistry", Dhanpat Rai Publishing, 2012.
- S.S. Dara, "A Textbook on experiment and calculation in engineering chemistry", S. Chand and Company, 9th revised edition, 2015.

PROGRAMMING FOR PROBLEM SOLVING LAB (Common to All Programs)

Instruction4 Periods per weekDuration of SEE3HoursSEE50MarksCIE50 Marks

Credits 2

Course Objectives: The objectives of this course are

- 1. Setting up programming environment.
- 2. Develop Programming skills to solve problems.
- 3. Use of appropriate C programming constructs to implement algorithms.
- 4. Identification and rectification of coding errors in program.
- 5. Develop applications in a modular fashion.
- 6. Manage data using files.

Course Outcomes: On Successful completion of the course, students will be able to

- 1. Identify and setup program development environment.
- 2. Design and test programs to solve mathematical and scientific problems.
- 3. Identify and rectify the syntax errors and debug program for semantic errors
- 4. Implement modular programs using functions.
- 5. Represent data in arrays, pointers, structures and manipulate them through a program.
- 6. Create, read, and write to and from simple text files.

Lab experiments

- 1. Familiarization with programming environment.
- 2. Simple computational problems using arithmetic expressions.
- 3. Problems involving if-then-else structures.
- 4. Iterative problems e.g., sum of series.
- 5. 1D Array manipulation.
- 6. 2D arrays and strings.
- 7. Matrix problems, String operations.
- 8. Simple functions.
- 9. Recursive functions.
- 10. Pointers and structures.
- 11. Dynamic memory allocation and error handling.
- 12. File handling:

Text Books:

- 1. Pradeep Dey and Manas Ghosh, "Programming in C", Oxford Press, 2nd Edition, 2017.
- 2. Reema Tharaja "Introduction to C Programming", Second Edition, OXFORD Press, 2015.

Online Resources:

- 1. https://www.tutorialspoint.com/cprogramming/index.htm
- 2. https://www.w3resource.com/c-programming/programming-in-c.php
- 3. https://www.w3schools.in/c-tutorial/

20ME C02

WORKSHOP / MANUFACTURING PRACTICE

Instruction 5P Hours per week
Duration of SEE 3Hours
SEE 50Marks
CIE 50Marks
Credits 2.5

Course Objectives:

- 1. Give a feel of Engineering Practices & develop holistic understanding of various Engineering materials and Manufacturing processes.
- 2. Develop skills of manufacturing, safety, precision, quality, intelligent effort, optimization, positive &team work attitude to get things right the first time.
- 3. To provide basic knowledge of Steel, Plastic, Composite and other materials for suitable applications.
- 4. Study of Principle and hands on practice on techniques of fabrication, welding, casting, manufacturing, metrology, and allied skills.
- 5. To advance important hard & pertinent soft skills, productivity, create skilled manpower which is cognizant of industrial workshop components and processes and can communicate their work in a technical, clear and effective way.

Course Outcomes: At the end of the course, the students are able to

- 1. Understand safety measures to be followed in workshop to avoid accidents.
- 2. Identify various tools used in fitting, carpentry, tin smithy, house wiring, welding, casting and machining processes.
- 3. Make a given model by using workshop trades including fitting, carpentry, tinsmithy and House wiring.
- 4. Perform various operations in welding, machining and casting processes.
- 5. Conceptualize and produce simple device/mechanism of their choice.

List of Exercises

CYCLE 1

Exercises in Carpentry

- 1. To plane the given wooden piece to required size
- 2. To make a lap joint on the given wooden piece according to the given dimensions.
- 3. To make a dove tail-joint on the given wooden piece according to the given dimensions.

Exercises in Tin Smithy

- 1. To make a rectangular box from the given sheet metal with base and top open. Solder the corners.
- 2. To make a scoop.
- 3. To make a pamphlet box.

Exercises in Fitting

- 1. To make a perfect rectangular MS flat and to do parallel cuts using Hacksaw
- 2. To make male and female fitting using MSflats-Assembly1
- 3. To make male and female fitting using MSflats-Assembly2

Exercises in House Wiring

- 1. Wiring of one light point controlled by one single pole switch, a three pin socket controlled by a single pole switch, and wiring of one buzzer controlled by a bellpush
- 2. Wiring of two light points connected in series and controlled by single pole switch. Verify the above circuit with different bulbs. Wiring of two light points connected in parallel from two single pole switches and a three pin socket
- 3. Stair case wiring-wiring of one light point controlled from two different places independently using two 2- ways witches.

CYCLE 2

Exercises in Casting

- 1. Study of Sand casting process and its applications.
- 2. Green sand moulding practice for a single piece pattern
- 3. Green sand moulding practice for a split pattern with a horizontal core

Exercises in Welding

- Study of gas welding equipment and process. Identification of flames, making of Butt joint with gas welding.
- 2. Study of Arc welding process, making Butt joint with DCSP, DCRP
- 3. Study of Arc welding process, making Lap joint with A.C

Exercises in Machine shop

- 1. Study of Machine Tools like Lathe, Drilling, Milling and Shaper.
- 2. Facing, Plain turning and Step turning operations on Lathe machine.
- 3. Knurling and Taper turning on Lathe machine

Open ended Exercise:

 Student should produce a component /mechanism by applying the knowledge of any one trade or combination of trades.

Text Books:

- 1. Hajra Choudhury S.K., Hajra Choudhury A.K. and Nirjhar Roy S.K., "Elements of Workshop Technology", Vol. I, 2008 and Vol. II, 2010, Media promoters and publishers private limited, Mumbai.
- 2. Kalpakjian S. And Steven S. Schmid, "Manufacturing Engineering and Technology", 4th edition, Pearson Education India Edition, 2002.
- 3. Rao P.N., "Manufacturing Technology", Vol. I and Vol. II, Tata McGraw Hill House, 2017.

- 1. Gowri P. Hariharan and A. Suresh Babu, "Manufacturing Technology I", Pearson Education, 2008.
- 2. Roy A. Lindberg, "Processes and Materials of Manufacture", 4th edition, Prentice Hall India, 1998.

20ME C03

ENGINEERING EXPLORATION

Instruction 4 Hours per week
Duration of SEE
Nil
SEE
Nil
CIE
50 Marks
Credits
1.5

Prerequisites: Nil

Course Outcomes: At the end of the course, the students are able to

- 1. Understand the role of an engineer as a problem solver.
- 2. Identify multi-disciplinary approaches in solving an engineering problem.
- 3. Build simple systems using engineering design process.
- 4. Analyze engineering solutions from ethical and sustainability perspectives.
- 5. Use basics of engineering project management skills in doing projects.

UNIT-I

Role of Engineers: Introduction, science, engineering, technology, engineer, scientist, role of engineer, various disciplines of engineering, misconception of engineering, expectations for the 21st century engineer and NBA graduate attributes.

Engineering problems and Design: Multidisciplinary facet of design, pair wise comparison chart, introduction to econometrics system, generation of multiple solution, Pugh chart, motor and battery sizing concepts, introduction to PCB design.

UNIT-II

Mechanisms: Basic components of a mechanism, degrees of freedom or mobility of a mechanism, 4-bar chain, crank rocker mechanism, slider crank mechanism, simple robotic arm building.

Platform-based development: Introduction to programming platforms (Arduino) and its essentials, sensors, transducers and actuators and their interfacing with Arduino.

UNIT-III

Data Acquisition and Analysis: Types of data, descriptive statistics techniques as applicable to different types of data, types of graphs and their applicability, usage of tools (MS-Office /Open Office/ Libre Office / Scilab) for descriptive statistics, data acquisition (temperature and humidity) using sensors interfaced with Arduino, exporting acquired data to spreadsheets, and analysis using representation.

UNIT-IV

Process Management: Introduction to Agile practice, significance of team work, importance of communication in engineering profession, project management tools, checklist, timeline, Gantt chart, significance of documentation.

UNIT -V

Engineering Ethics & Sustainability in Engineering: Identifying Engineering as a profession, significance of professional ethics, code of conduct for engineers, identifying ethical dimensions in different tasks of engineering, applying moral theories and codes of conduct for resolution of ethical dilemmas.

Sustainability in Engineering: Introduction, sustainability leadership, life cycle assessment, carbon foot print.

Text Books:

- 1. Clive L. Dym, Patric Little, Elizabeth J Orwin, "Engineering Design: A project-based introduction", 4th edition, Willey.
- 2. Matthew Python, "Arduino programming for beginners", Independently published, 2020.
- 3. Patrick F. Dunn, "Measurement and data Analysis for engineering and science", third edition, 2014.
- 4. Andrew Stellman, Jennifer Greene, "Head First Agile: A brain-friendly guide to Agile principles, ideas, and real-world practices", Kindle Edition.

- Charles B. Fleddermann, "Engineering ethics", fourth edition, Prentice Hall, 2012.

 Rob Lawlor, "Engineering in society", second edition, Royal academy of engineering.

 Richard Dodds, Roger Venables, "Engineering for sustainable development: Guiding principles", The Royal Academy of engineering, 2005. Richard S. Paul, "Robot Manipulators: Mathematics, Programming, and Control", MIT Press.
- 4.

S. No	Name of the module	Work Hours	Marks	Evaluation		
1	Role of Engineers	4	-			
2	Engineering Design	16	5	Evaluation - I		
3	Mechanisms	6	3			
4	Engineering Ethics	2	2			
5	Platform-based Development	16	5	Evaluation - II		
6	Data Acquisition and Analysis	6	4			
7	Project Management	4	4	Evaluation-III		
8	Sustainability in Engineering	6	2			
9	Course Project Reviews	12	20			
10	Code of conduct	-	5	Final Evaluatio		
	Total	72	50			



CHAITANYA BHARATHI INSTITUTE OF TECHNOLOGY (AUTONOMOUS)

Scheme of Instructions of II Semester of B.E. –Civil Engineering as per AICTE Model Curriculum 2020-21

DEPARTMENT OF CIVIL ENGINEERING

SEMESTER-II

G	Course Code	Title of the Course	Scheme of Instruction			Scheme of			
S. No			Hours per Week			Duration of SEE	Maximum Marks		Credits
			L	T	P/D	in Hours	CIE	SEE	
		TI	HEORY	7					
1	20MT C06	Vector Calculus and Differential Equations	3	1	-	3	40	60	4
2	20EG C01	English	2	-	-	3	40	60	2
3	20PY C05	Mechanics and Materials Science	3	ı	-	3	40	60	3
4	20CE C02	Engineering Mechanics – II	3	-	-	3	40	60	3
5	20EE C01	Basic Electrical Engineering	3	-	-	3	40	60	3
PRACTICAL									
6	20EG C02	English lab	-	-	2	3	50	50	1
7	20PY C08	Mechanics and Materials Science Lab	-	-	4	3	50	50	2
8	20EE C02	Basic Electrical Engineering Lab	-	-	2	3	50	50	1
9	9 20MB C02 Community Engagement			eld + 2	2P/W	-	50	-	1.5
	TOTAL				8	-	400	450	20.5

L: Lecture T: Tutorial P: Practical

CIE - Continuous Internal Evaluation SEE - Semester End Examination

20MT C06

VECTOR CALCULUS AND DIFFERENTIAL EQUATIONS (Common to ECE, EEE, MECH, CHEM, CIVIL)

Instruction 3 L+1T /2P Hours per week
Duration of SEE 3 Hours
SEE 60 Marks
CIE 40 Marks
Credits 4

Course Objectives:

- 1. To explain double and triple integrals of various functions and their significance.
- 2. To explain Scalar and vector functions with its Physical interpretations.
- 3. To discuss vector line, surface and volume integrals.
- 4. To explain relevant methods to solve first order differential equations.
- 5. To discuss the solution of higher order differential equations.

Course Outcomes:

Upon completing this course, students will be able to:

- 1. Calculate the areas and volumes.
- 2. Apply the vector differential operators to Scalars and Vector functions.
- 3. Solve line, surface & volume integrals by Greens, Gauss and Stoke's theorems.
- 4. Calculate the solutions of first order linear differential equations.
- 5. Solve higher order linear differential equations.

IINIT_I

Multivariable Calculus (Integration): Applications of definite integrals to evaluate surface areas and volumes of revolutions. Double integrals, Change of order of integration, Area enclosed by plane curves, Triple integrals, Volumes of solids.

UNIT-II

Vector Differential Calculus: Scalar and vector point functions, Gradient, Directional derivative, potential function. Divergence, Physical interpretation of Divergence, Curl, Physical interpretation of curl, vector identities.

UNIT-III

Vector Integral Calculus: Line integral, Surface integral and Volume integral, Green's theorem in a plane, Stoke's theorem and Gauss's divergence theorem (without proof). Fluid flow, Physical interpretation of the divergence.

UNIT-IV

First Order Ordinary Differential Equations: Exact differential equations, Equations reducible to exact equations, Linear equations, Bernoulli's equation, Clairaut's equation, Riccati's equation, Orthogonal trajectories, Chemical reactions and solutions, Rate of decay of Radio-active materials.

UNIT-V

Higher Orders Linear Differential Equations: Higher order linear differential equations with constant coefficients, rules for finding Complementary function, Particular Integral and General solution. Method of variation of parameters, solution of Euler-Cauchy equation. Ordinary point, singular point, regular singular point and Power Series solution. Applications: LR and LCR circuits.

Text Books:

- 1. B.S.Grewal, Higher Engineering Mathematics, Khanna Publishers, 44th Edition, 2017.
- 2. Erwin Kreyszig, Advanced Engineering Mathematics, 9th Edition, John Wiley & Sons, 2006.

- 1. N.P.Bali and Dr. Manish Goyal, A text book of Engineering Mathematics, Laxmi Publications, 9th edition, 2014.
- 2. R.K.Jain, S.R.K.Iyengar, Advanced Engineering Mathematics, Narosa Publications, 5th edition, 2016.
- 3. G.B. Thomas and R.L. Finney, Calculus and Analytic Geometry, 9th Edition, Pearson, Reprint, 2002

ENGLISH

(Common to all branches)

Instruction2 Hours per weekDuration of SEE3 HoursSEE60 MarksCIE40 Marks

Credits 2

Course Objectives: This course will introduce the students:

- 1. To the role and importance of communication while developing their basic communication skills in English.
- 2. To basics of writing coherent paragraphs and formal emails.
- 3. To techniques of writing a précis and formal letters by using acceptable grammar and appropriate vocabulary.
- 4. To description, definition and classification of processes while enabling them to draft formal reports following a proper structure.
- 5. To gaining adequate reading comprehension techniques.

Course Outcomes: After successful completion of the course the students will be able to:

- 1. Illustrate the nature, process and types of communication and communicate effectively without barriers.
- 2. Construct and compose coherent paragraphs, emails and adhering to appropriate mobile etiquette.
- 3. Apply techniques of precision to write a précis and formal letters by using acceptable grammar and appropriate vocabulary.
- 4. Distinguish formal from informal reports and demonstrate advanced writing skills by drafting formal reports.
- 5. Critique passages by applying effective reading techniques

UNIT-I Understanding Communication in English:

Introduction, nature and importance of communication; Process of communication; Types of communication - verbal and non-verbal; Barriers to communication; Intrapersonal and interpersonal communication; Understanding Johari Window.

Vocabulary & Grammar: The concept of Word Formation; Use of appropriate prepositions and articles.

UNIT-II Developing Writing Skills I:

Paragraph writing. – Structure and features of a paragraph; Cohesion and coherence. Rearranging jumbled sentences. Email and Mobile etiquette.

Vocabulary & Grammar: Use of cohesive devices and correct punctuation.

UNIT-III Developing Writing Skills II:

Précis Writing; Techniques of writing precisely. Letter Writing – Structure, format of a formal letter; Letter of request and the response

Vocabulary and Grammar: Subject-verb agreement.

Use of prefixes and suffixes to form derivatives. Avoiding redundancies.

UNIT-IV Developing Writing Skills III:

Report writing – Importance, structure, elements of style of formal reports; Writing a formal report. **Vocabulary and Grammar:** Avoiding ambiguity - Misplaced modifiers. Use of synonyms and antonyms.

UNIT-V Developing Reading Skills:

The reading process, purpose, different kinds of texts; Reading comprehension; Techniques of comprehension – skimming, scanning, drawing inferences and conclusions.

Vocabulary and Grammar: Words often confused; Use of standard abbreviations.

Text Books:

- 1. Language and Life: A Skills Approach, Board of Editors, Orient Black Swan, 2017. 2. Swan Michael, Practical English Usage.OUP. 1995.

- 1. Wood F.T, Remedial English Grammar, Macmillan, 2007
- 2. Zinsser William, On Writing Well, Harper Resource Book, 2001
- 3. Sanjay Kumar and PushpLata, Communication Skills. Oxford University Press, 2011.

20PY C05

MECHANICS AND MATERIALS SCIENCE (Common to Civil & Mechanical)

Instruction 3 L / week
Duration of SEE 3 Hours
SEE 60 Marks
CIE 40 Marks
Credits 3

Course Objectives: The objectives of the course is to make the student

- 1. Acquire knowledge about physics of oscillations and rotational motion
- 2. Understand the physical properties of crystalline and magnetic materials
- 3. Aware of characteristic properties of dielectric materials and superconductors
- 4. Familiarize with coherent properties of light waves.

Course Outcomes: At the end of the course, the student will be able to

- 1. Compare the various types of oscillations
- 2. Demonstrate rotational motion of rigid body
- 3. Classify different types of crystals and their imperfections
- 4. Identify magnetic and dielectric materials for engineering applications
- 5. Make use of lasers and superconductors in technological applications

UNIT-I

Oscillations: Simple harmonic motion—Harmonic oscillator—Damped harmonic motion — over damped, critically damped and under damped oscillators—Forced oscillations and resonance.

UNIT-II

Rigid body Dynamics: Definition of rigid body–Rotational kinematic relations–Angular momentum and torque–Equation of motion for a rotating rigid body–Inertia tensor and its properties– Euler's equations and applications: law of energy conservation and law of conservation of angular momentum.

UNIT-III

Crystallography: Space lattice –Unit cell –Crystal systems –Bravais lattices –Number of atoms per unit cell – Coordination number –Atomic radius –Packing fraction (for *sc*, *bcc*, *fcc*) –Lattice planes – Miller indices – Bragg's law –Experimental determination of lattice constant of a cubic crystal by powder X-ray diffraction method–Structure of NaCl.

Crystal Imperfections: Classification of defects –Point defects –Concentration of Schottky and Frenkel defects.

UNIT-IV

Dielectric Materials: Introduction –Dielectric polarization –Types of dielectric polarization: electronic & ionic polarizations (quantitative); orientation & space-charge polarizations (qualitative) –Frequency and temperature dependence of dielectric polarization –Determination of dielectric constant (Schering bridge method) –Ferroelectricity–Barium titanate–Applications of ferroelectrics.

Magnetic Materials: Origin of magnetism — Magnetic moment - Bohr magneton—Classification of magnetic materials: dia, para, ferro, anti-ferro and ferrimagnetic materials — Weiss molecular field theory — Domain theory — Hysteresis curve, soft and hard magnetic materials — Applications.

UNIT-V

Lasers: Characteristics of lasers – Einstein's coefficients – Amplification of light by population inversion - Ruby, He-Ne, semiconductor laser – Applications of lasers in engineering and medicine.

Fiber Optics: Introduction – Construction – Principle – Propagation of light through an optical fiber – Numerical aperture and acceptance angle – Step-index and graded-index fibers – Pulse dispersion – Fiber losses – Fiber optic communication system – Applications

Superconductors: General properties of superconductors – Meissner's effect – Type I and Type II superconductors – BCS theory (qualitative) – Applications.

Text Books:

- 1. B.K. Pandey and S. Chaturvedi, *Engineering Physics*, Cengage Publications, 2012.
- 2. M.N. Avadhanulu and P.G. Kshirsagar, *A Text Book of Engineering Physics*, S. Chand Publications, 2014.
- 3. M. Arumugam, *Materials Science*, Anuradha Publications, 2015.
- 4. S.L. Gupta and Sanjeev Gupta, *Modern Engineering Physics*, Dhanpat Rai Publications, 2011.

- 1. R. Murugeshan and Kiruthiga Sivaprasath, *Modern Physics*, S. Chand Publications S. Chand Publications, 2014.
- 2. V. Rajendran, Engineering Physics, McGraw-Hill Education Publications, 2013.
- 3. P.K. Palanisamy, *Engineering Physics*, Scitech Publications, 2012.
- 4. V. Raghavan, *Materials Science and Engineering*, Prentice Hall India Learning Private Limited; 6th Revised edition, 2015.

20CE C02

ENGINEERING MECHANICS - II

Instruction: 3 L Hours per week
Duration of SEE: 3 Hours
SEE: 60 Marks
CIE: 40 Marks
Credits: 3

Course Outcomes: At the end of the course, the student will be able to

- 1) determine moment of inertia for plane areas and mass moment of inertia of bodies.
- 2) analyse the rectilinear and curvilinear translations of particles.
- 3) solve kinetics problems for particles and connected bodies using dynamic equilibrium equations.
- 4) apply work-energy principle for particles and connected bodies.
- 5) apply linear impulse momentum principle for the motion of bodies and understand concept of simple harmonic motion and its applications

UNIT - I

Moment of Inertia (M.I):

Area Moment of Inertia: Introduction, polar moment of inertia, radius of gyration and transfer formula. Moment of Inertia of plane areas by integration, M.I of composite areas, product of inertia, transfer formula for product of inertia.

Mass Moment of Inertia: Mass Moment of Inertia of thin plates, radius of gyration, transfer formula and mass moment of inertia of cylinder, sphere, cone and composite bodies.

UNIT - II

Kinematics: Rectilinear and curvilinear translation (uniform and variable acceleration). Rectangular, normal and tangential components of acceleration.

UNIT - III

Kinetics: General Principles of kinetics, D' Alembert's principle and its application to particle motion, angle of banking and connected bodies.

UNIT - IV

Work-Energy Method: Work done by a force and kinetic energy of a particle. Equation of work energy for translation - applied to particle motion and connected bodies.

UNIT - V

Impulse and Momentum: Introduction to linear impulse-momentum, principle of conservation of linear momentum and its applications. Elastic impact and coefficient of restitution.

Mechanical Vibrations: Introduction, amplitude, time period, frequency, simple harmonic motion, free vibrations, simple pendulum.

Text Books:

- 1. K. Vijaya Kumar Reddy and J. Suresh Kumar, "Singer's Engineering Mechanics: Statics and Dynamics", B. S. Publications (SI Units), 3rd edn. Rpt., 2019.
- 2. A. Nelson., "Engineering Mechanics", Tata McGraw Hill, Delhi, 2010.

- 1. Irving H. Shames, "Engineering Mechanics", 4th Edition, Prentice Hall, 2006.
- 2. R. C. Hibbeler, "Engineering Mechanics: Principles of Statics and Dynamics", Pearson Press, 2006.
- 3. Basudeb Bhattacharyya, "Engineering Mechanics", Oxford University Press, 2nd edn., 2016.

BASIC ELECTRICALENGINEERING

Instruction 3 Hours per week
Duration of SEE 3 Hours
SEE 60 Marks
CIE 40 Marks
Credits 3

Course Objectives:

- 1. To understand the behaviour of different circuit elements R, L & C, and the basic concepts of electrical AC circuit analysis
- 2. To understand the basic principle of operation of AC and DC machines
- 3. To know about different types of electrical wires and cables, domestic and industrial wiring. safety rules and methods of earthing

Course Outcomes: After the completion of this course, the student will be able to

- 1. Understand the concepts of Kirchhoff's laws and to apply them in superposition, Thevenin's and Norton's theorems to get the solution of simple dc circuits
- 2. Obtain the steady state response of RLC circuits with AC input and to acquire the basics, relationship between voltage and current in three phasecircuits.
- 3. Understand the principle of operation, the emf and torque equations and classification of AC and DC machines
- Explain various tests and speed control methods to determine the characteristic of DC and AC machines.
- 5. Acquire the knowledge of electrical wiring, types of wires, cables used and Electrical safety precautions to be followed in electrical installations.
- 6. Recognize importance of earthing, methods of earthing and various low-tension switchgear used in electrical installations

UNIT-I

DC Circuits: Electrical circuit elements (R, L and C), voltage and current sources, Kirchhoff current and voltage laws, analysis of simple circuits with dc excitation, Superposition, Thevenin and Norton Theorems, Time-domain analysis of first-order RL and RC circuits.

UNIT-II

AC Circuits: Representation of sinusoidal waveforms, peak and rms values, phasor representation, real power, reactive power, apparent power, power factor, Analysis of single-phase ac circuits consisting of R, L, C, RL, RC. Three phase balanced circuits, voltage and current relations in star and delta connections.

UNIT-III

Transformers: Construction, Working principle, EMF Equation, Ideal and Practical transformer, Equivalent circuit of Transformer, OC and SC tests on a transformer, Efficiency and Regulation

UNIT-IV

DC and AC Machines: DC Generators: Construction, Principle of operation, EMF equation, Classification, Characteristics of shunt, series and compound generators.

DC Motors: Classification, Torque equation, Characteristics, Efficiency, Speed Control of Series and Shunt Motors.

Three - Phase Induction Motors: Principle of operation, Applications,

UNIT-V

Electrical Installations: Electrical Wiring: Types of wires and cables, Electrical Safety precautions in handling electrical appliances, electric shock, first aid for electric shock, safety rules.

Components of LT Switchgear: Switch Fuse Unit (SFU), MCB, ELCB, Earthing (Elementary Treatment only), Elementary calculations for energy consumption

Text Books:

- 1. L. S. Bobrow, Fundamentals of Electrical Engineering, Oxford University Press,2011.
- 2. E. Hughes, Electrical and Electronics Technology, Pearson, 2010.

- 1. D. P. Kothari & I. J. Nagrath, "Basic Electrical Engineering", Tata McGraw Hill, 2010.
- V. D. Toro, "Electrical Engineering Fundamentals", Prentice Hall India,1989
 D.C. Kulshreshtha, "Basic Electrical Engineering", McGrawHill,2009
- 4. P.V. Prasad, S. sivanagaraju, R. Prasad, "Basic Electrical and Electronics Engineering" Cengage Learning, 1st Edition,2013

ENGLISH LAB

(Common to all branches)

Instruction2 Hours per weekDuration of SEE3 HoursSEE50 MarksCIE50 Marks

Credits 1

Course Objectives: This course will introduce the students:

- 1. To nuances of Phonetics and give them sufficient practice in correct pronunciation.
- 2. To word stress and intonation.
- 3. To IELTS and TOEFL material for honing their listening skills.
- 4. To activities enabling them overcome their inhibitions while speaking in English with the focus being on fluency rather than accuracy.
- 5. To team work, role behavior while developing their ability to discuss in groups and making oral presentations.

Course Outcomes: After successful completion of the course the students will be able to:

- 1. Define the speech sounds in English and understand the nuances of pronunciation in English
- 2. Apply stress correctly and speak with the proper tone, intonation and rhythm.
- 3. Analyze IELTS and TOEFL listening comprehension texts to enhance their listening skills.
- 4. Determine the context and speak appropriately in various situations.
- 5. Design and present effective posters while working in teams, and discuss and participate in Group discussions.

Exercises

- 1. **Introduction to English Phonetics**: Introduction to auditory, acoustic and articulatory phonetics, organs of speech: the respiratory, articulatory and phonatory systems.
- 2. **Sound system of English**: Phonetic sounds and phonemic sounds, introduction to international phonetic alphabet, classification and description of English phonemic sounds, minimal pairs. The syllable: types of syllables, consonant clusters.
- 3. Word stress: Primary stress, secondary stress, functional stress, rules of word stress.
- 4. **Rhythm & Intonation**: Introduction to Rhythm and Intonation. Major patterns, intonation of English with the semantic implications.
- 5. **Listening skills** Practice with IELTS and TOEFL material
- 6. **Public speaking** Speaking with confidence and clarity in different contexts on various issues.
- 7. **Group Discussions -** Dynamics of a group discussion, group discussion techniques, body language.
- 8. **Pictionary** weaving an imaginative story around a given picture.
- 9. **Information Gap Activity** Writing a brief report on a newspaper headline by building on the hints given
- 10. **Poster presentation** Theme, poster preparation, team work and presentation.

- 1. T Balasubramanian. A Textbook of English Phonetics for Indian Students, Macmillan, 2008.
- 2. J Sethi et al. A Practical Course in English Pronunciation (with CD), Prentice Hall India, 2005.
- 3. Priyadarshi Patnaik. Group Discussions and Interviews, Cambridge University Press Pvt Ltd., 2011
- 4. Aruna Koneru, Professional Speaking Skills, Oxford University Press, 2016

MECHANICS AND MATERIALS SCIENCELAB (Common to Civil & Mechanical)

Instruction 4 Periods / week
Duration of SEE 3 Hours
SEE 50 Marks
CIE 50 Marks
Credits 2

Course Objectives:

The objectives of the course is to make the student

- 1. Apply the concepts of physics while doing experiments
- 2. Learn the working of lasers and optical fibers
- 3. Understand the properties of magnetic and dielectric materials
- 4. Capable of measuring mechanical properties of solids and liquids
- 5. Understand the motion electrons in electric and magnetic fields

Course Outcomes:

At the end of the course, the student will be able to

- 1. Estimate the error in an experimental measurement
- 2. Make use of lasers and optical fibers in engineering applications
- 3. Recall the physical properties of dielectrics and magnetic materials
- 4. Find the mechanical properties of solids and viscosity of liquids
- 5. Demonstrate the motion of electrons in electric and magnetic fields

Experiments

1.	Error Analysis	:	Estimation of errors in the determination of time period of a torsional pendulum
2.	Flywheel	:	Determination of moment of inertia of given flywheel
3.	Compound Pendulum	:	Determination of acceleration due to gravity
4.	Young's Modulus	:	Determination of Young's modulus of the given steel bar/wooden scale
			by non-uniform bending method
5.	Helmholtz's Resonator	:	Determination of resonating volume of air and neck correction
6.	Melde's Experiment -	:	Determination of frequency of the electrically maintained vibrating bar/fork
7.	Viscosity of Liquid	:	Determination of viscosity of a given liquid by oscillating disc method
8.	Coupled Oscillator	:	To determine the coupling constant of a coupled oscillator performing parallel and antiparallel oscillation
9.	Dielectric Constant	:	Determination of dielectric constant of given PZT sample
10.	M & H Values	:	Determination of magnetic moment M of a bar magnet and absolute value H of horizontal component of earth's magnetic field
11.	B-H Curve	:	Determination of hysteresis loss of given specimen
12.	Thermoelectric Power	:	Determination of thermoelectric power of given sample
13.	Laser	:	Determination of wavelength of given semiconductor laser
14.	Optical Fiber	:	Determination of numerical aperture and power losses of given optical fiber
15.	e/m of an electron	:	Determination of specific charge of an electron by J.J. Thomson

NOTE: A minimum of TWELVE experiments should be conducted.

method

20EEC02

BASIC ELECTRICAL ENGINEERING LAB

Instruction 2 Hours per week
Duration of SEE 3 Hours
SEE 50 Marks
CIE 50 Marks
Credits 1

Course Objectives:

- 1. To acquire the knowledge of different types of electrical elements and to verify the basic electrical circuit laws and theorems.
- 2. To determine the parameters and power factor of a coil, calculate the time and frequency responses of RLC circuits and to familiarize with measurement of electric power & energy.
- 3. To determine the characteristics of Transformers, dc, ac machines and switchgear components

Course Outcomes: At the end of the course, the students are expected to

- 1. Get an exposure to common electrical components, their ratings and basic electrical measuring equipment.
- 2. Make electrical connections by wires of appropriate ratings and able to measure electric power and energy.
- 3. Comprehend the circuit analysis techniques using various circuital laws and theorems.
- 4. Determine the parameters of the given coil and calculate the time response of RL & RC series circuits.
- 5. Recognize the basic characteristics of transformer and components of switchgear.
- 6. Understand the basic characteristics of dc and ac machine by conducting different types of tests on them.

List of Laboratory Experiments/Demonstrations:

- 1. Demonstration of Measuring Instruments and Electrical Lab components.
- 2. Verification of KCL and KVL.
- 3. Time response of RL and RC series circuits.
- 4. Determination of parameters of a choke or coil by Wattmeter Method
- 5. Verification of Thevenin's and Norton's theorems
- 6. Turns ratio /voltage ratio verification of single phase Transformers
- 7. Open Circuit and Short Circuit tests on a given single phase Transformer
- 8. Observation of Excitation Phenomenon in Transformer
- 9. Measurement of three phase power in a balanced system using two Wattmeter method.
- 10. Measurement of 3-Ph Energy by an Energy Meter (Demonstration of Principle)
- 11. Load test on DC Shunt motor
- 12. Speed control of DC Shunt motor
- 13. Demonstration of Low Tension Switchgear Equipment/Components
- 14. Demonstration of cut out section of Machines like DC Machine, Induction Machine etc.

Note: TEN experiments to be conducted from the above list.

20MBC02

COMMUNITY ENGAGEMENT

Instruction 3 Hours per week (30 hours

field work & 2 hours per week)

SEE Nil
CIE 50 Marks
Credits 1.5 Credits

Course Objectives: The main Objectives of this Course are to:

- 1. Develop an appreciation of Rural culture, life-style and wisdom among the Students.
- 2. Learn about the various livelihood activities that contribute to Rural economy.
- 3. Familiarize the Rural Institutions and the Rural Development Programmes in India.

Course Outcomes: After the completion of this Course, Student will be able to:

- 2. Gain an understanding of Rural life, Culture and Social realities.
- 3. Develop a sense of empathy and bonds of mutuality with Local Communities.
- 4. Appreciate significant contributions of Local communities to Indian Society and Economy.
- 5. Exhibit the knowledge of Rural Institutions and contributing to Community's Socio-Economic improvements.
- 6. Utilise the opportunities provided by Rural Development Programmes.

Module I Appreciation of Rural Society

Rural life style, Rural society, Caste and Gender relations, Rural values with respect to Community, Nature and Resources, elaboration of 'soul of India lies in villages' (Gandhi), Rural Infrastructure.

Module II Understanding Rural Economy and Livelihood

Agriculture, Farming, Landownership, Water management, Animal Husbandry, Non-farm Livelihood and Artisans, Rural Entrepreneurs, Rural markets, Rural Credit Societies, Farmer Production Organization/Company.

Module III Rural Institutions

Traditional Rural organizations, Self-Help Groups, Panchayati Raj Institutions (Gram Sabha), Gram Panchayat, Standing Committees, Local Civil Society, Local Administration.

Module IV Rural Development Programmes

History of Rural Development in India, Current National Programmes: Sarva Shiksha Abhiyan, Beti Bhachao, Beti Padhao, Ayushman, Bharat, Swachh Bharat, PM AwasYojana, Skill India, Gram Panchayat Decentralised Planning, NRLM, MNREGA etc.

Text Books:

- Singh, Katar, Rural Development: Principles, Policies and Management, Sage Publications, New Delhi, 2015.
- 2. A Hand book on Village Panchayat Administration, Rajiv Gandhi Chair for Panchayati Raj Studies, 2002.
- 3. United Nations, Sustainable Development Goals, 2015, un.org/sdgs
- 4. M.P Boraia, Best Practices in Rural Development, Shanlax Publishers, 2016.

Journals:

- 1. Journal of Rural development (published by NIRD & PR, Hyderabad).
- 2. Indian Journal of Social Work, (by TISS, Bombay).
- 3. Indian Journal of Extension Educations (by Indian Society of Extension Education).
- 4. Journal of Extension Education (by Extension Education Society).
- 5. Kurukshetra (Ministry of Rural Development, GOI).
- 6. Yojana (Ministry of Information & Broadcasting, GOI).